





S. Schulz

page has recently published his 10th article in Angewandte Chemie in the last 10 years: "A Gallium-Substituted Distibene and an Antimony-

The author presented on this

Distibene and an Antimony-Analogue Bicyclo-[1.1.0]butane: Synthesis and Solid-State Structures": L. Tuscher, C. Ganesamoorthy, D. Bläser, C. Wölper, S. Schulz, Angew. Chem. Int. Ed. 2015, 54, 10657; Angew. Chem. 2015, 127, 10803.

Stephan Schulz

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Education: 1992 Undergraduate degree, University of Göttingen

1994 PhD supervised by Herbert W. Roesky, University of Göttingen 1994–1996 Postdoctoral position with Richard F. Jordan, University of Iowa 2001 Habilitation in the group of Edgar Niecke, University of Bonn

Awards: 1994 Research Fellowship (Deutsche Forschungsgemeinschaft; DFG); 1996 Liebig Fellowship

(Fonds der Chemischen Industrie); 1999 Fellowship for habilitation (DFG)

Current research Structure and reactivity of (low-valent) organometallic compounds and their application in

interests: materials synthesis

Hobbies: Jogging, watching soccer, spending time with friends and family

The most important thing I learned from my students is to be prepared for everything.

The principal aspect of my personality is to tell things as they are.

The natural talent I would like to be gifted with is not to tell things as they are.

If I could be a piece of lab equipment, I would be an NMR tube since I like to spin around.

My motto is "Maach et joot, ävver nit zo off" (a saying from the Rhineland meaning "do it well, but not too often", i.e., quality over quantity).

am waiting for the day when someone will realize that scientific discoveries not necessarily follow three-year research programs.

Last time I went to the pub my friends and I solved almost every problem in the world—unfortunately we couldn't remember the next day.

If I could be anyone for a day, I would be food taster for Keith Richards (but only for one day).

admire people with a consistent life plan who always know exactly what comes next.

advise my students to explore their own way in life, and to be careful with advice given by people like me.

The secret of being a successful scientist is something I am still looking for.

My favorite structure is that of chlorine azide—a beautiful beast.

My 5 top papers:

- 1. "Van der Waals epitaxial MOCVD-growth of $(Bi_xSb_{1-x})_2Te_3$ (0 < x < 1) films": G. Bendt, J. Sonntag, A. Lorke, W. Assenmacher, U. Hagemann, S. Schulz, *Semicond. Sci. Technol.* **2015**, *30*, 085021. (Chemistry plays an essential role in materials science.)
- "Temperature-Dependent Electron Shuffle in Molecular Group 13/15 Intermetallic Complexes": C. Ganesamoorthy, D. Bläser, C. Wölper, S. Schulz, Angew. Chem. Int. Ed. 2014, 53, 11587; Angew. Chem. 2014, 126, 11771. (Understanding electronic interactions is the most important thing in chemistry.)
- "A Comparison of the Solid-State Structures of Halogen Azides XN₃ (X = Cl, Br, I)": B. Lyhs, D. Bläser, C. Wölper, S. Schulz, G. Jansen, *Angew. Chem. Int. Ed.* 2012, 51, 12859; *Angew. Chem.* 2012, 124, 13031. (The

- compounds show remarkably different intermolecular interactions in the solid state.)
- "Synthesis of Hexagonal Sb₂Te₃ Nanoplates by Thermal Decomposition of the Single-Source Precursor (Et₂Sb)₂Te": S. Schulz, S. Heimann, J. Friedrich, M. Engenhorst, G. Schierning, W. Assenmacher, *Chem. Mater.* 2012, 24, 2228. (Illustrates the great potential of single-source precursors in materials synthesis.)
- 5. "Structural Characterization of a Base-Stabilized [Zn.]²⁺ Cation": S. Schulz, D. Schuchmann, I. Krossing, D. Himmel, D. Bläser, R. Boese, *Angew. Chem. Int. Ed.* 2009, 48, 5748; *Angew. Chem.* 2009, 121, 5859. (The [Zn.]²⁺ dication became accessible only when using the principle of base stabilization.)

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